

Low Molecular Weight Zinc-Dependent Acid Phosphatase from Chicken's Heart

ASMA SAEED*, A. H. SHAH, AND A. R. KHAN

Department of Biological Sciences, Gomal University, D. I. Khan, Pakistan.

(Received 14th November 2006, revised 9th December 2006)

Summary: Zn⁺⁺-dependent acid phosphatase from chicken's heart was partially purified by ammonium sulfate precipitation, heat treatment at 60 °C, CM-Cellulose chromatography and gel filtration on Sephadex G-100 to the specific activity of 0.8 $\mu\text{mol min}^{-1} \text{mg}^{-1}$ at pH 6.0. A 34-folds purification was achieved with the recovery of 8 %. The enzyme had molecular weight 57 kDa as revealed by gel filtration and 29 kDa by SDS-polyacrylamide gel electrophoresis, indicating the dimeric nature of the enzyme. The enzyme was activated by Zn⁺⁺, Co⁺⁺ and Mn⁺⁺ and inhibited by phosphate, while EDTA, tartrate and flouride had little or no effect on the activity, which are the potent inhibitors of high molecular weight acid phosphatases.

Introduction

Phosphatases belong to the hydrolase class and catalyze the hydrolysis of almost any phosphate monoesters. They play a very important role in biological systems, which are linked to processing involving energy metabolism, metabolic regulation and cellular signal transduction pathways [1]. Phosphatases are widely distributed in nature and differ greatly in their sizes, structures and mechanisms. They can be classified in several ways. In one classification, phosphatases have been divided into five classes, namely alkaline phosphatases, purple acid phosphatases, high molecular weight acid phosphatases, low molecular weight acid phosphatases and protein phosphatases.

In vertebrate tissues, at least two classes of acid phosphatases (E.C.3.1.3.2), differing in molecular weight and localization have within the cells have been described [2]. Another class of acid phosphatase, which requires Zn⁺⁺ for activity called Zn⁺⁺- dependent acid phosphatase has been reported [3].

Phosphatase also exists in two major forms, which have similar biochemical properties but vary in molecular weight and tissue distribution [4]. High molecular weight Zn⁺⁺- dependent acid phosphatase (~ 100 kDa) is expressed in the liver of animal tissues [3, 5], while low molecular weight Zn⁺⁺ - dependent acid phosphatase is about 57-62 kD and has been found in other tissues like brain, heart, skeletal muscle, erythrocytes, lungs, spleen and stomach [6,

7]. Small intestine contains significant quantities of both forms [8, 9].

Purification and biochemical properties of chicken's liver high having molecular weight Zn⁺⁺- dependent acid phosphatase were reported recently [10].

Contrarily, in this paper, we report the existence of low molecular weight Zn⁺⁺ - dependent acid phosphatase in the chicken's heart. The enzyme was purified using almost the similar procedure to that reported for liver enzyme and characterized with respect to molecular weight, metal ion dependence, inhibition, pH and temperature optima and kinetic properties.

Results and Discussion

Low molecular weight Zn⁺⁺ - dependent acid phosphatase (57-63 kDa) was observed to be present in the chicken's heart. Purification procedure for this enzyme was the same as that for high molecular weight enzyme of chicken liver [10]. The major purification steps are summarized in Table-1. Elution profile for CM-Cellulose chromatography and Sephadex G-100 is shown in Figs. 1 and 2. A-34 fold purification was achieved with the recovery of 8 %. The enzyme had specific activity of 0.8 $\mu\text{mol min}^{-1} \text{mg}^{-1}$ at pH 6.0 at 37° C. SDS-polyacrylamide gel electrophoresis (12 %) showed almost a single band at 29 kDa with Coomassive Blue Staining (Fig. 3)

*To whom all correspondence should be addressed.

Table 1- Partial purification scheme for 57 kDa Zn⁺⁺-dependent acid phosphatase from 50 gm of Chicken's heart

Steps	Vol. (ml)	Act. (U/ml)	T.Act. (U)	Prot. (mg/ml)	S.A (U/mg)	P.F	Rec. %
Extract	188	0.26	48.88	11	0.023	1	100
60% ammonium sulphate precipitation	22	0.943	20.7	35	0.026	1.17	42.4
Heat treatment at 60° C and dialysis	30	0.22	6.6	5.34	0.041	2	13.4
CM-Cellulose Chromatography at pH 5.5	3	1.53	4.6	11.7	0.131	5.7	9.4
Sephadex G-100	17.5	0.23	4.02	0.29	0.795	34.57	8.2

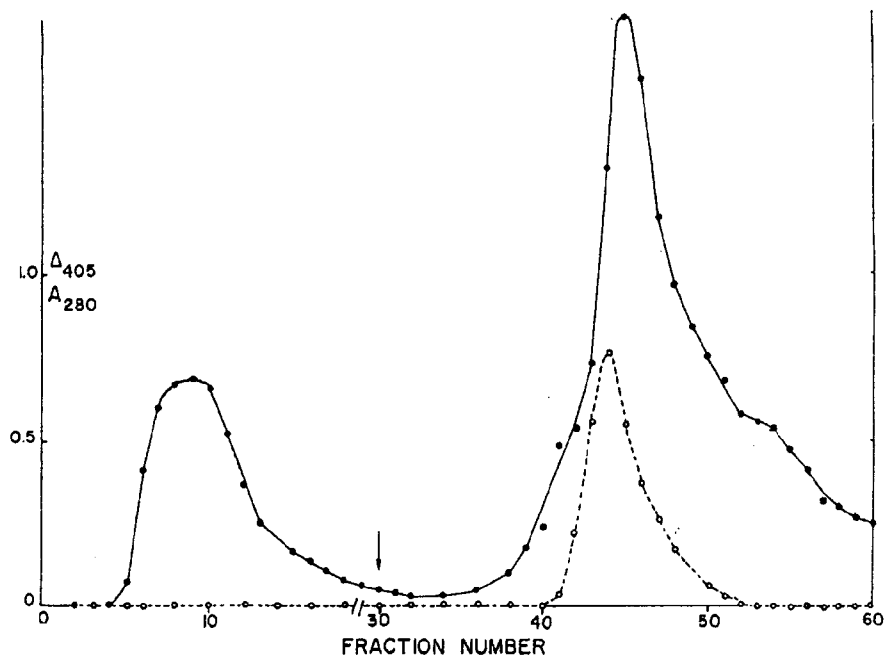


Fig. 1: CM-cellulose chromatography at pH 5.5. 30 ml sample (7U) was applied, Column (1.5x20 cm), flow rate 30 ml/h, each fraction 3-4 ml. Arrow indicates starting point of gradients. Ordinate: Protein at 280 nm (● ●), activity at 405 nm (o.....o).

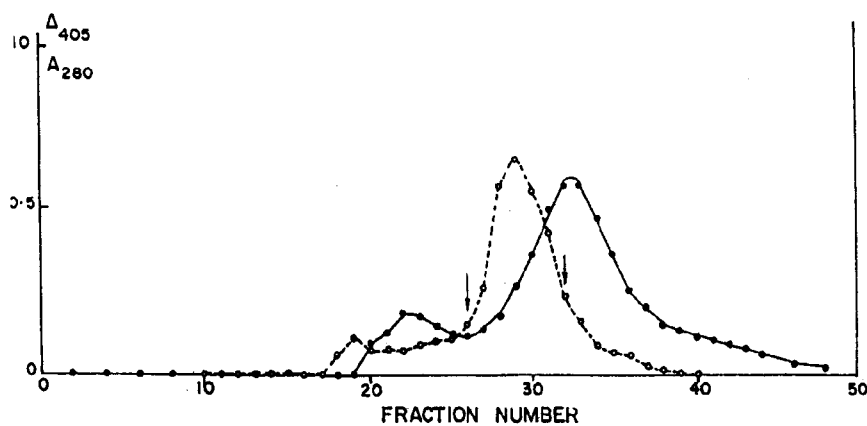


Fig. 2: Gel chromatography on Sephadex G-100. 3 ml sample (6U) was applied, column (1.8x85 cm), flow rate 20 ml/h, each fraction 3.5 ml. Ordinates: Protein at 280 nm (●—●) activity at 405 nm (o.....o)

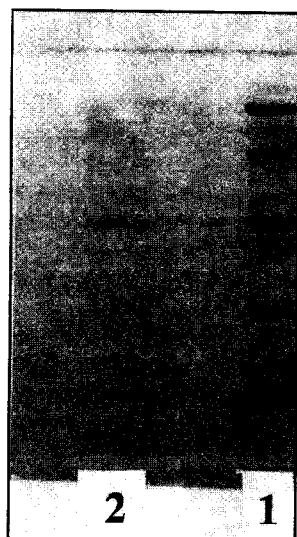


Fig.3 SDS-Polyacrylamide gel electrophoresis of Zn⁺⁺ - dependent acid phosphatase.

Lane 1: Standard proteins used from top to bottom were Albumin Bovine (66,000); Oval albumin (45,000); Carbonic anhydrase (29,000); Trypsin inhibitor (20,000); Lact albumin (14,200).

Lane 2 5 ul of Zn⁺⁺- dependent acid phosphatase.

after rechromatography on Sephadex G-100. The molecular weight of native enzyme obtained by gel filtration on Sephadex G-100 was estimated to be 57 kDa (Fig. 4), which indicated this enzyme to be a dimer. 57 kDa enzyme is also present in rat brain, testis, prostate and adrenal glands [6], which is distinguished from the 100 kDa enzyme isolated from the liver of various vertebrates [11-13]. High molecular weight acid phosphate and low molecular weight acid phosphate from chicken's heart displayed molecular weight values (Fig. 4 b) being similar to those previously reported for the high and low molecular weight enzymes isolated from other sources [13].

Chicken heart low molecular weight Zn⁺⁺ - dependent acid phosphatase activity showed maximum activity over pH of 4.5-6.0 (Fig. 5). The optimal temperature for activity was 65° C (Fig. 6). Km was found to be 0.6mM.

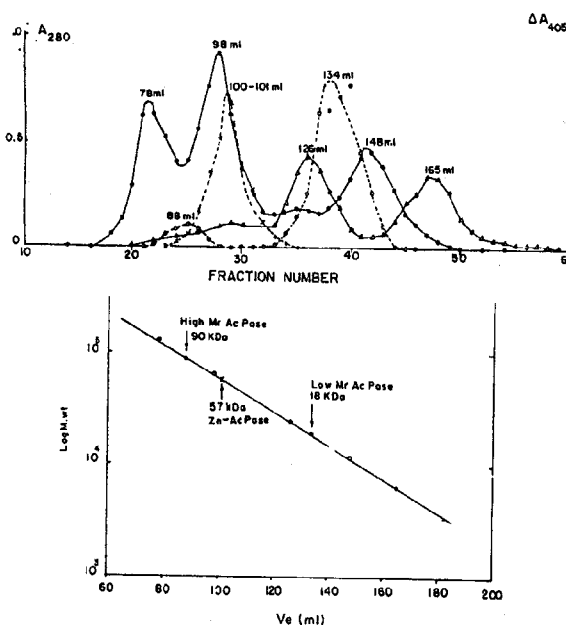


Fig. 4: (a) Elution pattern of standard proteins on Sephadex G-100.

The column (1.8x85 cm) was equilibrated and eluted with 0.01M acetate buffer pH 5 containing 0.1 M sodium chloride at flow rate of 20 ml/h and 3.5 ml fractions were collected; o 72 ml, Blue dextran 2000:Boine serum albumin (66,000), e 98 ml (●—●); carbonic anhydrase (29,000), Ve 126 ml (Δ—Δ); cytochrome C(12,400), Ve 148 ml (●—●); Aprotinin (6,500), Ve 165 ml (Δ—Δ) Zn⁺⁺ pendent Acid phosphatase, Ve 100 m (x....x); High Molecular weight acid phosphatase, Ve 88 ml (o.....o) lecular weight acid phosphatase, e 134 ml (o.....).

(b) Linear graph of log M.W. versus elution volumes of standard proteins.

The low molecular weight Zn⁺⁺ - dependent acid phosphatase showed no p-nitrophenyl phosphate hydrolyzing activity in the absence of Zn⁺⁺ ion. The hydrolysis was observed in the presence of Zn⁺⁺, Mn⁺⁺ and Co⁺⁺. Zn⁺⁺ was the most effective cation for activating the catalytic activity of the enzyme (Fig. 7). 5mM Zn⁺⁺ concentration was required for maximum hydrolysis. Similar finding was obtained with high molecular weight liver enzyme. In comparison with Zn⁺⁺, Mn⁺⁺ and Co⁺⁺ showed 16 % and 12 % activation respectively. Mg⁺⁺ and other divalent ions such as Ca⁺⁺, Fe⁺⁺ a Ni⁺⁺ had little or no effect (Table 2). These results are in good agreement

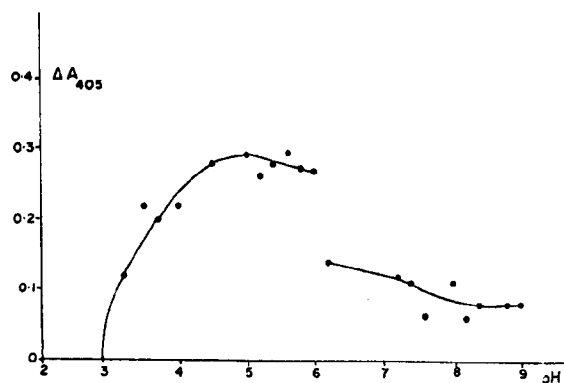


Fig. 5: Optimum pH of acid phosphatase from chicken heart.

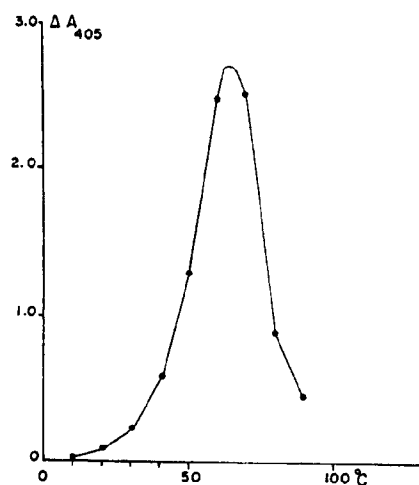


Fig. 6: Optimum temperature of Zn²⁺-dependent acid phosphatase activity.

Table 2- Effect of metal ions as activators (5mM) on the p-nitrophenyl Phosphate hydrolysis

Metal ion	Δ A ₄₀₅	Activity (%)
Water	0.007	1.34
ZnCl ₂	0.521	100
MnCl ₂	0.083	16
CoCl ₂	0.062	12
MgCl ₂	0.005	1
CaCl ₂	0.00	0
NiCl ₂	0.00	0
FeSO ₄	0.0	0

with that of high molecular weight enzymes of chicken liver, bovine liver and human liver. In another experiment, the effect of divalent cations at a concentration of 5mM on Zn²⁺-dependent acid phosphates activity is reported in Table 3, which

showed that Mn²⁺ appreciably altered the Zn²⁺ dependent acid phosphatase activity whereas Ba²⁺, Co²⁺, Ca²⁺, Mg²⁺ and Ni²⁺ did not, while heavy

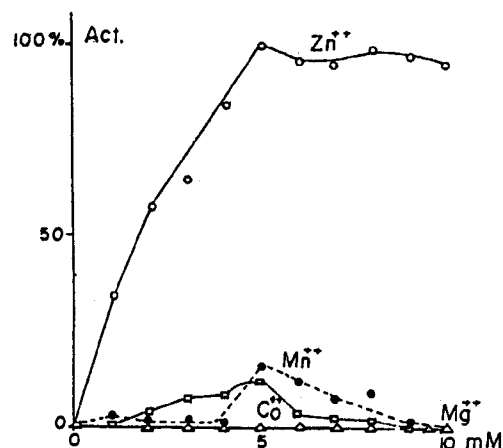


Fig. 7: Influence of metal ions as activators on the hydrolysis of p-nitrophenyl phosphate. The activity was determined in the presence of metal ions at concentrations in abscissa. The maximum activity was expressed by taking the highest activity in any fraction as 100% and the other activities were expressed as a percentage of the maximum activity.

Table 3- Effect of metal ions (5mM) on Zn²⁺-dependent acid phosphatase activity

Divalent metal ion	Activity (%)
None	100
MnCl ₂	37.11
BaCl ₂	100
CoCl ₂	90.9
CaCl ₂	100
MgCl ₂	81.03
NiCl ₂	85.5
FeCl ₃	0
FeSO ₄	0
HgCl ₂	14.9

Table 4- Effect of inhibitors on Zn²⁺-dependent acid phosphatase activity

Concen- tration	Tartrate (%)	EDTA (%)	Phosphate (%)	Fluoride (%)
	Activity	Activity	Activity	Activity
0 mM	100	100	100	100
1 mM	100	-	98	100
2 mM	100	-	77	94.3
3 mM	90	-	-	95.5
4 mM	87	98	68	98
5 mM	80.6	-	65	-
6 mM	79	96	60	-
8 mM	77	95	52	100
10 mM	69	93	37	93.3

metals such as Fe^{++} , Cu^{++} and Hg^{++} inhibited the enzyme strongly.

The effect of various inhibitors or modifiers on the activity of Zn^{++} - dependent acid phosphatase was tested by adding tartrate, fluoride, EDTA and Na_3PO_4 at concentrations ranging from 0 to 10mM. Table 4 shows that fluoride, tartrate and EDTA had little or no effect on the activity, which are potent inhibitors of high molecular weight acid phosphatase. Phosphate indeed seemed to be a strong inhibitor. The inhibition of this enzyme was greater than that of liver enzyme [10]. The Zn^{++} dependence, pH activity curve and modifiers sensitivity strongly suggested that the 57 kDa Zn^{++} - dependent acid phosphatase displays biochemical properties very similar to those of 100 kDa enzymes but with a molecular weight, $a =$ which is half that of liver enzyme. Both liver and heart enzymes are composed of two apparently identical subunits of about 50 kDa and 29 kDa respectively, reflecting molecular and structural analogies. Thus, purification of 57 kDa Zn^{++} - dependent acid phosphatase is necessary to elucidate these points.

Experimental

Material

Chicken's heart was obtained from a local poultry shop. Para-nitrophenyl phosphate, molecular weight markers, CM-cellulose and Sephadex G-100 were purchased from Sigma Chemical Co. The material for SDS-polyacrylamide gel electrophoresis was obtained from Bio-Rad and Across Chemical Co. All other reagents were of the highest purity commercially available.

Enzyme Assays

The Zn^{++} - dependent acid phosphatase activity was determined as described by Panara [14] at 37 °C using 4mM p-nitrophenyl phosphate in 125 mM acetate buffer at pH 6.0, containing 5mM ZnCl_2 and 10 mM NaF and 50-100 μl of enzyme solution in a final volume of 1ml. After 5 minute, the reaction was stopped by the addition of 1 ml of 0.1 N KOH and the absorbance was measured at 405 nm ($\epsilon = 1.8 \times 10^4 \text{ M}^{-1}\text{cm}^{-1}$). Units of activity were expressed as μ mol of p-nitrophenol released per minute. The specific activity was defined as units per mg of protein.

The activity over a pH range of 3.2-9.0, the temperature optima, determined at 10°C intervals from 10-90°C and the effect of metal ions were determined as reported earlier [11].

Protein Determination

Protein concentration was estimated by the Biuret method [15]. For column effluents, the relative protein concentration was estimated from absorbance at 280 nm.

Molecular Weight Determination

The molecular weight of Zn^{++} - dependent acid phosphatase and acid phosphatases was determined by gel filtration on a Sephadex G-100 column (1.8 x 85 cm), which was previously calibrated with the following standard proteins: aprotinin (6,500), cytochrome C (12,400), carbonic anhydrase (29,000) and bovine serum albumin (66,000).

SDS polyacrylamide gel electrophoresis was carried out by the method of Laemmli [16].

Enzyme Purification

Step 1: Isolation of low molecular weight Zn^{++} - dependent acid phosphatase:

The fresh chicken's heart was washed with 10mM Tris-HCl buffer at pH 7.4, containing 0.25 M sucrose and 1 mM EDTA and was cut into small pieces. The pieces were homogenized gently in a blender containing small volume of 10 mM acetate buffer at pH 6.0 containing 1mM EDTA, 2mM β -mercaptoethanol and 0.1 mM PMSF. The homogenate was adjusted to 3ml buffer per gram of original tissue. The homogenate was centrifuged at 3840 x g for 30 minutes and the supernatant was collected and precipitated by 60 % saturation ammonium sulfate. The precipitate collected by centrifugation at 3840 x g was dissolved in extraction buffer.

Step 2: Heat Treatment

The supernatant was heated at 60 °C for 2-5 minutes and centrifuged at 10,000 x g for 20 minutes. The pellet was discarded and the supernatant was collected. The sample was dialyzed against 0.01 M acetate buffer at pH 5.5, containing 0.1 mM PMSF, 2 mM β -mercaptoethanol overnight with 2-3 times changes of buffer.

Step 3: CM-Cellulose Chromatography at pH 5.5

Dialyzed sample was centrifuged at 3840 x g for 30 min. and the clear supernatant was applied to CM-Cellulose column (3 x 10cm) equilibrated with dialyzing buffer. The column was washed with the same buffer to remove unbound proteins. Linear gradient from 0-0.5M NaCl in the same buffer (total 300 ml) was applied to elute bound enzyme as shown in Fig. 1. The most active fractions were pooled and precipitated by adding solid ammonium sulfate to 70 % saturation. The precipitate was collected by centrifugation at 10,000 x g for 30 min. and dissolved in a small amount of 0.01M acetate buffer at pH 6.0 containing 1mM EDTA and 0.1 M PMSF.

Step 4: Gel Chromatography on Sephadex G-100

The enzyme was applied to Sephadex G-150 column (1.8 x 70cm), equilibrated and eluted with 0.01M acetate buffer at pH 6.0 containing 1mM EDTA, 2mM β -mercaptoethanol and 0.1M NaCl. The elution profile is shown in Fig. 2. The active fractions were pooled and the enzyme was concentrated in dialyzing tube over which sucrose was sprinkled for 4 to 5 h.

Acknowledgments

This work was supported in part by a grant from Gomal University Research Project No 12/GU and PSF project No R and G/F-GU/Chem (173).

References

1. A. W. Murray, *Nature*, **359**, 599 (1992).
2. J. B. Vincent, M. W. Crowder and B. A. Averill, *Trends Biochem. Sci.*, **17**, 105 (1992).
3. A. Angiolillo and F. Panara, *Biol. Pharm. Bull.*, **20**(12), 1235 (1997).
4. A. Caselli, P. Cirri, S. Bonifacio, G. Manao, G. Camici, G. Cappugi, G. Moneti and G. Ramponi, *Biochimica et Biophysica Acta*, **1290**, 241 (1996).
5. F. Panara, A. Angiolillo, A. Fagotti, I. Di Rosa and R. Pascolini, *Trends Comp. Biochem. Physiol.*, **1**, 675 (1993).
6. F. Panara, A. Angiolillo, A. Fagotti, I. Di Rosa, F. Simoncelli and R. Pascolini, *Int. J. Biochem.*, **24**(10), 1619 (1992).
7. S. Fujimoto, H. Gotoh, T. Ohbayashi, H. Yazaki and A. Ohara, *Biol. Pharm. Bull.*, **16**(8), 745 (1993).
8. S. Fujimoto, I. Okano, Y. Tanaka, Y. Sumida, J. Tsuda, N. Kawakami and S. Shimohama, *Biol. Pharm. Bul.*, **19**(6), 882 (1996).
9. J. Tsuda, T. Kimura, H. Tanino, S. Shimohama and S. Fujimoto, *Biol. Pharm. Bull.*, **21**(11), 1218 (1998).
10. Asma Saeed, S. A. Malik and A. Saeed, *Jour. Chem. Soc. Pak.*, **24**(3), 215 (2002).
11. F. Panara, *Biochem. J.*, **235**, 265 (1986).
12. F. Panara, *Int. J. Biochem.*, **20**(4), 457 (1988).
13. F. Panara, N. Massetti, A. Angiolillo, A. Fagotti and R. Pascolini, *J. Exp. Zoo.*, **254**, 119 (1990).
14. F. Panara, *Journal of Fish Biology*, **51**, 275 (1997).
15. G. Beisenherz, H. J. Boltze, T. Bucher, R. Czok, K. H. Garbade, E. Meyer-Arendt and G. Pfeleiderer, *Z. Naturforsch.*, **B 8**, 555 (1953).
16. U. K. Laemmli, *Nature*, **227**, 680 (1970).