High Performance Liquid Chromatographic Separation of Putrescine and Cadaverine using 2-Hydroxynaphthaldehyde as a Derivatizing Reagent

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(Received 11th December, 1995, revised 15th May, 1996)

Summary: Biological active diamines, cadaverine and putrescine together with 1,3-propylenediamine and 1,2-propylenediamine react with 2-hydroxynaphthaldehyde to form bis (2-hydroxynaphthaldehyde) 1,5-diiminopentane (HN₂CA), bis(2-hydroxynaphthaldehyde) 1,4-diiminopropane (HN₂Pu) bis (2-hydroxynaphthaldehyde) 1,3-diiminopropane (HN₂Pn). The derivatives are observed to be fluorescent in visible region. The wavelength of excitation (Ex) and emission (Em) were determined for optimal fluorescence of derivatives. The linear calibration for each of the diamine is observed at 0.32-1.6 μg/ml. The diamine derivatives HN₂Pu, HN₂Pn also separated on Nova Pak C-18 column when eluted with methanol. The spectrofluorimetric detection was obtained at 450 nm using Ex at 315 nm.

Introduction

The naturally occurring diamines 1.3diaminopropane, 1,4-diaminobutane (Putrescine) 1,5-diaminopentane (Cadaverine) have been shown to be present ubiquitously in procarytotes and eucaryotes [1-4]. In mammals including man, putrescine is generally present in low concentration except in tissues that are growth stimulated or have a high cell proliferative compartment such as bone marrow [5]. Putrescine in mammalian cells is formed by direct decarboxylation of orinithine [1], The interests in the determination of diamines have increased since Russell reported that patients with matastic cancer excreted increased levels of putrescine in urine [6]. A number of studies have been carried out to determine the concentration of diamines and polyamines in biological fluids and tissues, the variations in concentrations have been determined in normal, tumor affected tissues and

after treatment with different anticancer agents. A number of different methods have also been reported for the determination of the diamines and polyamines, including spectrophotometry [7], spectrofluorimetry, thin layer chromatography [8], ion, exchange chromatography [9], gas chromatography [10] and liquid chromatography [11,12]. The liquid chromatographic procedure involves mostly minhydrin [8] o-phthaldehyde [13], 9fluorenylmethyl chloroformate [14], N-succinimidyl-3-ferrocenylpropinate [15], benzoyl-chloride [16], 3,5-dinitrobenzovlchloride [17], densyl chloride [18], fluorescamine [19] acetyl-acetone [20] 8-hydroxyquinoline sulphonic acid [21] and 2-(1-pyrenyl)ethyl chloroformate [22] as reagents for post column and precolumn derivatization. Spectrophotometric, spectrofluorimetric or electrochemical detection is conveniently obtained.

Recently 2-hydroxy-1-naphthaldehyde (NH) has been used as a derivatizing reagent for putrescine (Pu), cadaverine (CA), 1,2-diaminopropane (pn) and 1,3-diaminopropane (PR) for HPLC separation and determination of the diamines [23]. Spectrophotometric detection was obtained at 260 nm. In the present work the diamine derivatives, HN₂Pu, HN₂CA, HN₂Pn and HN₂PR (Fig. 1) have been characterized on spectrofluorimetry and their HPLC separation has been examined using spectrofluorimetric detection.

Fig. 1: Structural diagrams diamines derivatives.

Results and Discussion

The potentials of the diamine derivatives HN₂Pn, HN₂PR, HN₂Pu and HN₂CA, conveniently prepared by simple condensation were investigated for the fluorometric determination of diamines. The absorption spectrum of each of the derivative was recorded in methanol. Different wavelength of maximum absorbances were fixed as excitation emission wavelengths and wavelength scanned. The wavelength at which maximum fluorescence intensity was observed was selected. Again the emission wavelength was fixed and excitation wavelength was scanned and optimized excitation wavelength was confirmed. The results of this study are summarized in Table-1. Linear calibrations were obtained with 0.32-1.6 µg/ml by intensity against fluorescence plotting concentration with coefficient of correlation (y) of 0.999, 0.982, 0.9987 and 0.999 for HN2Pn, HN2PR, HN₂Pu, and HN₂CA respectively, HN₂Pu, has highest sensitivity followed by in sequence HN₂Pn, HN2PR, HN2CA (Fig. 2).

Table Optimized conditions for fluorometric determination of diamine derivatives

	Compound	Excitation wavelength	Emission wavelength
1.	Bis(2-hydroxy-1-naphthaldehyde- 1,2-diiminopropane	240	350
2.	Bis(2-hydroxy-1-naphthaldehyde)	247	450
3.	1,3-diiminopropane Bis(2-hydroxy-1-naphthaldehyde)-	310	460
4.	1,4-diiminobutane Bis(2-hydroxy-1-naphthaldehyde)-	395	450
	1,5-diiminopentane		

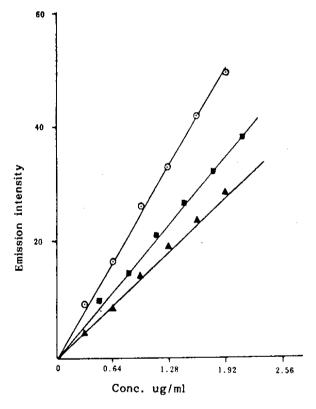
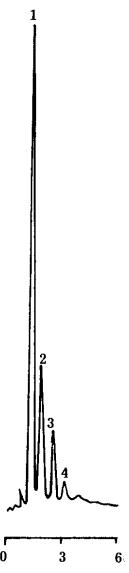


Fig. 2: Calibration plate of 1) HN₂Pu 2) HN₂PR and 3) HN₂CA

The diamine derivatives indicated a reasonable sensitivity to enable to determine the diamine at sub µg/ml, however the derivatizing reagent HN also indicates some fluorescence at the wavelengths selected for HN₂PR, HN₂Pn, HN₂Pu, and HN₂CA and would interfere when present in excess, added as derivatizing agent. It was therefore HPLC coupled with spectrofluorimetric detector was used for the separation of HN, HN₂PR, HN₂Pu and HN₂CA. All of the compounds were

easily eluted and separated from the reversed phase C-18 column when eluted with methanol, using flow rate of 0.9 mL/min. The detection was obtained at 450 nm (Em) using 315 nm as excitation wavelength (Fig. 3). However, HN₂Pn, coeluted with HN₂PR and failed to separate using the described conditions.



Ret: Time in min.

Fig. 3: HPLC separation of 1) NH 2) PR 3) Pu 4)
CA as derivatives of NH.
Column Nova Pak C-18, 5μ (150 x 3.9 mm id). Elution with methanol 0.9 mL/min. Detection Spectrofluorometeric EX 315 nm, and Em 45 nm.

When the derivatization procedure was used HN completely separated form diamine derivatives and did not interfere with the determination of the diamines. Linear calibration curves for each of the PR, Pu and CA at the optimized wavelengths for the fluorimetric determination were obtained with 0.4-2.5 µg/mL, 2-6 µg/mL 0.8-6 µg/mL respectively. The detection limits measured at least three times the background noice were obtained 40 ng/mL, 200 ng/mL, 80 ng/mL for PR, Pu and CA respectively.

Experimental

Bis(2-hydroxynaphthaldehyde) 1,2-diiminopropane (HN₂Pn), bis (2-hydroxynaphthaldehyde) 1,3-diiminopropane (HN₂PR), bis (2-hydroxynaphthaldehyde) 1,4-diiminobutane (HN₂Pu) and bis (2hydroxynaphthaldehyde) 1,5-diiminopentane (HN₂CA) were prepared by condensation of HN with Pn, PR, Pu or CA in 2:1 molar ratio in ethanol as reported [23]. Fluoresence measurements were carried out on Hitachi F-1200 spectrofluorimeter. Absorption spectra of the compounds were recorded on Hitachi 220 spectrophotometer. Hitachi 655A liquid chromatograph connected with Hitachi F-1200 spectrofluorimeter, Rheodyne 7125 injector, and Hitachi 561 recorder was used. Separations were made on a Nova Pak C-18 (150 x 3.9 mm id) (Millipore Ltd. USA) column.

Derivatization procedure

Solution (1-5 mL) containing (PR, Pu, CA) diamines (0-150 μ g) each was added sodium bicarbonate buffer pH 8 (2 ml) and HN (5 mL) (1 mg/ml in methanol). The mixture was warmed at 60°C for 5 min. and volume was adjusted to 25 mL with methanol. The solution (1 μ l) was injected on HPLC column and solutes were eluted with methanol 0.9 mL/min and spectrofluorimetric detection was made at Ex:315 and EM; 450 mm.

References

- 1. O. Heby, Differentiation, 18, 1 (1981)
- H. Guy Williams-Ashman and Zoe N. Canellakis, Spring 421 (1979).
- 3. H. Taber and C.M. Tabor, *Pharmacol. Rev.*, 16, 245 (1964).

- M. Stepita-Klauco and H. Dolezalova, *Nature*, 252, 158 (1974).
- D.H. Russell and B.G.M. Durie, Polyamines as Biochemical Markers of Normal and Malignant Growth, Raven Press, New York (1978).
- 6. D.H. Russell, Nature, 233, 144 (1971).
- 7. J.P. Gouygou, C. Martin, C. Singuin and P. Dur Oceanis, 15, 599 (1989).
- 8. S. Bardocz, T. Karsai and P. Elodi, *Chromatographia*, 20,23 (1989).
- 9. R.C. Adlaka and V.R. Villanveva, J. Chromatogr., 187, 422 (1980).
- 10. X. Jiang, Biomid. Chromatogr., 4, 73 (1990).
- 11. L. Gerbant, Clin. Chem., 37, 2117 (1991).
- 12. S. Moret, R. Bortotomeazzi, G. Lercker, J. Chromatogr., 591, 175 (1992).
- 13. J.L. Corbin, H.B. March and G.A. Peters, *Plant Physiol.*, **96**, 434 (1989).
- R.S. Gillbert, G.G., Gonzalez, Leo III Hawell and Criag V. Bycis, Anal. Biochem., 199, 86

- (1991).
- K. Shimada, T.Oe and M. Tanaka, J. Chromatogr., 487, 247 (1989).
- 16. G. Table, M.R. Schiavo, J. Chromatogr., Biomed. Appl., 614, 153 (1993).
- 17. S. Wongyai, P. Oefner and G. Bonn, *Biomed. Chromatogr.*, 2, 254 (1988).
- 18. P.M. Kabra, H.R. Lee, W.P. Lubich and L.J. Marton, *J. Chromatogr.*, **380**, 19 (1986).
- 19. M. Kai, T. Ogata, K. Haraguchi and O. Koichi, J. Chromatogr., 163, 151 (1979).
- 20. Y. Nishikawa, J. Chromatogr., 392, 349 (1987)
- 21. I. Pigulla and E. Roeder, Fresenius Z. Anal. Chem., 293, 404 (1978).
- 22. M.A. Cichy, D.L. Stegmeir, H. Veening, H.D. Becker, J. Chromatogr., Biomed. Appl. 613, 15 (1993).
- 23. M.Y. Khuhawar, M.B. Rind and G.A. Qureshi, J. Chem. Soc. Pak., 18, 119 (1996).